

Professor: J.D. WilsonTime available: 50 minsValue: 20%

No formula sheets or cell phones. **Formulae and data provided at the back.** Please record your multichoice answers on the scantron sheet, choosing the correct or most logical answers (incorrect options *may* feature false or misleading statements or ‘facts’). Respond to short answer and calculation questions in the space provided: **add your name**, tear off and submit. (You may keep other pages of this exam.)

A. Multi-choice (15 x 2/3 → 10 %)

1. Molecules of CO₂ combine a carbon atom (atomic number 12) with two oxygen atoms (oxygen having atomic number 16). Given that the universal gas constant is $R^* = 8.3143 \text{ J mol}^{-1} \text{ K}^{-1}$, which value below best approximates the specific gas constant R for CO₂?
 - (a) $190 \text{ J kg}^{-1} \text{ K}^{-1}$
 - (b) $0.044 \text{ kg mol}^{-1}$
 - (c) $287 \text{ J kg}^{-1} \text{ K}^{-1}$
 - (d) $0.190 \text{ J kg}^{-1} \text{ K}^{-1}$
 - (e) $300 \text{ J kg}^{-1} \text{ K}^{-1}$
2. Suppose the atmosphere of a certain planet were composed entirely of N₂, for which the specific gas constant is $R = 297 \text{ J kg}^{-1} \text{ K}^{-1}$. If the surface pressure and temperature at a certain point on this planet were (80 kPa, 50°C) then what was the surface density ρ ?
 - (a) 0.1 kg m^{-3}
 - (b) 0.5 kg m^{-3}
 - (c) 0.83 kg m^{-3}
 - (d) 1 kg m^{-3}
 - (e) 5.4 kg m^{-3}
3. Isolines of which “secondary field” are plotted on the CMC isobaric charts for the 850 hPa level?
 - (a) MSLP
 - (b) isobaric height
 - (c) isohumes (isolines of relative humidity)
 - (d) 1000-500 hPa thickness (DZ)
 - (e) temperature

4. Suppose 0.5 kg of dry air is contained in a rigid volume of 2 cubic metres. Which value best approximates the change in temperature (final minus initial) if 10^4 J of energy are added to this system?
- (a) -30K
 - (b) -15K
 - (c) $+15\text{K}$
 - (d) $+30\text{K}$
 - (e) $+60\text{K}$
5. At the time of the southern hemisphere summer solstice, what is the noon solar elevation at 23.5°N latitude?
- (a) 0°
 - (b) 23.5°
 - (c) 43°
 - (d) 66.5°
 - (e) 90°
6. Defining the volumetric sensible heat content of a parcel of air as $h = \rho c_p T$ [J m^{-3}] and taking the ρc_p product as 10^3 [$\text{J m}^{-3} \text{K}^{-1}$], what is the value of the convective heat flux density Q_x if the x -component of the wind is $u = 5 \text{ m s}^{-1}$ and the temperature of the parcel is 300 K?
- (a) $1.5 \times 10^6 \text{ W m}^{-2}$
 - (b) $3 \times 10^5 \text{ K}$
 - (c) 120 W m^{-3}
 - (d) $120 \text{ W m}^{-2} \text{ s}^{-1}$
 - (e) 1500 J m^{-3}
7. Taking the conductivity of glass as $k = 1$ [$\text{W m}^{-1} \text{K}^{-1}$], what is the rate of conductive heat transfer across a window pane of area 1 m^2 and thickness $\Delta x = 2.5 \text{ mm}$ when the temperature difference across the pane is 4K ?
- (a) 1600 J
 - (b) 1600 W
 - (c) 160 W s^{-1}
 - (d) 100 W
 - (e) 10 W
8. A parcel of dry air at the surface ($z = 0 \text{ m}$) has a temperature of 20°C . It is lifted adiabatically to $z = 700 \text{ m}$ then sinks adiabatically to $z = 500 \text{ m}$. Its final temperature is?
- (a) 29°C
 - (b) 27°C
 - (c) 25°C
 - (d) 20°C
 - (e) 15°C

9. If a certain body has longwave emissivity $\epsilon = 0.96$ and its temperature is $T = 18^\circ\text{C}$, which answer below most closely states the wavelength λ_{max} of the peak in its thermal emission spectrum and its full-spectrum rate of emission E of longwave radiation?
- (a) $10\ \mu\text{m}$, $6 \times 10^{-3}\ \text{W m}^{-2}$
 - (b) $10\ \mu\text{m}$, $390\ \text{W m}^{-2}$
 - (c) $100\ \mu\text{m}$, $390\ \text{W m}^{-2}$
 - (d) $1\ \mu\text{m}$, $410\ \text{W m}^{-2}$
 - (e) $0.1\ \mu\text{m}$, $410\ \text{W m}^{-2}$
10. Suppose a parcel of air at the surface had state $(P_1, T_1) = (930\ \text{hPa}, 12^\circ\text{C})$. What is the parcel's temperature after it has been lifted adiabatically to the 850 hPa level? (Hint: Poisson's equation).
- (a) 11.7°C
 - (b) 12.3°C
 - (c) 278 K
 - (d) 293 K
11. Suppose on a sunny summer afternoon the net radiation over a flat field of bare soil were $Q^* = 500\ \text{W m}^{-2}$, and the sensible and latent heat fluxes were $Q_H = 180$, $Q_E = 300\ \text{W m}^{-2}$. What was the soil heat flux Q_G ?
- (a) $980\ \text{W m}^{-2}$
 - (b) $480\ \text{W m}^{-2}$
 - (c) $120\ \text{W m}^{-2}$
 - (d) $20\ \text{W m}^{-2}$
 - (e) $-980\ \text{W m}^{-2}$

For the remaining questions, please refer to Figures (1 – 2).

12. What was MSLP and what were the present weather conditions at the station closest to the **X** on Figure 1?
- (a) 915.6 hPa; overcast, rain, light ENE wind
 - (b) 915.6 hPa; overcast, snow, light WSW wind
 - (c) 1015.6 hPa; overcast, rain, light WSW wind
 - (d) 1015.6 hPa; overcast, snow, light WSW wind
 - (e) 1015.6 hPa; overcast, snow, light ENE wind

13. The wind observations on Figure 1 show anti-clockwise winds about the low pressure system near Churchill (Manitoba), with a component of cross-isobar flow towards low pressure. Which statement is **false**?
- (a) this is the normal pattern of flow about a N. hemisphere low
 - (b) the cross-isobar wind implies ascending vertical motion, possibly resulting in cloud and precip
 - (c) on the west side of the low, cold air is moving southward to replace milder air (cold advection)
 - (d) on the east side of the low, cold air is moving northward to replace milder air (warm advection)
 - (e) MSLP at the centre of the low was lower than 992 hPa
14. According to Figure 2, the 850 hPa analysis, at which point or points was cold advection occurring?
- (a) A
 - (b) A, B
 - (c) C
 - (d) B, D
 - (e) B
15. From Figure 2, how would you describe the 850 hPa wind over Edmonton?
- (a) NNW, over 15 m s^{-1} , parallel to height contours
 - (b) SSE, over 15 m s^{-1} , parallel to height contours
 - (c) NNW, over 30 m s^{-1} , perpendicular to height contours
 - (d) NNW, over 15 m s^{-1} , crossing height contours

Continue to Calculations on next page.

B. Computations (2 x 2 → 4 %)

Round your answers to three significant digits (e.g. 1.23 or 0.0123 or 1.23×10^{-2}), and state the units.

The hypsometric equation reads

$$\Delta z = z_2 - z_1 = \left[\frac{R_d \bar{T}_v}{g} \right] \ln \frac{P_1}{P_2},$$

where $R_d = 287 \text{ [J kg}^{-1} \text{ K}^{-1}]$ and $g = 9.81 \text{ [m s}^{-2}]$. Note that if $z_2 > z_1$, then $P_2 < P_1$.

B1. What is the thickness Δz of the 1000 hPa to 500 hPa atmospheric layer if the average virtual temperature of the layer is $\bar{T}_v = 0^\circ\text{C}$?

B2. Suppose the distance from ground to the 850 hPa isobaric surface was $\Delta z = 900 \text{ m}$ and that the average virtual temperature of that layer was $\bar{T}_v = 10^\circ\text{C}$? What is the pressure at the surface?

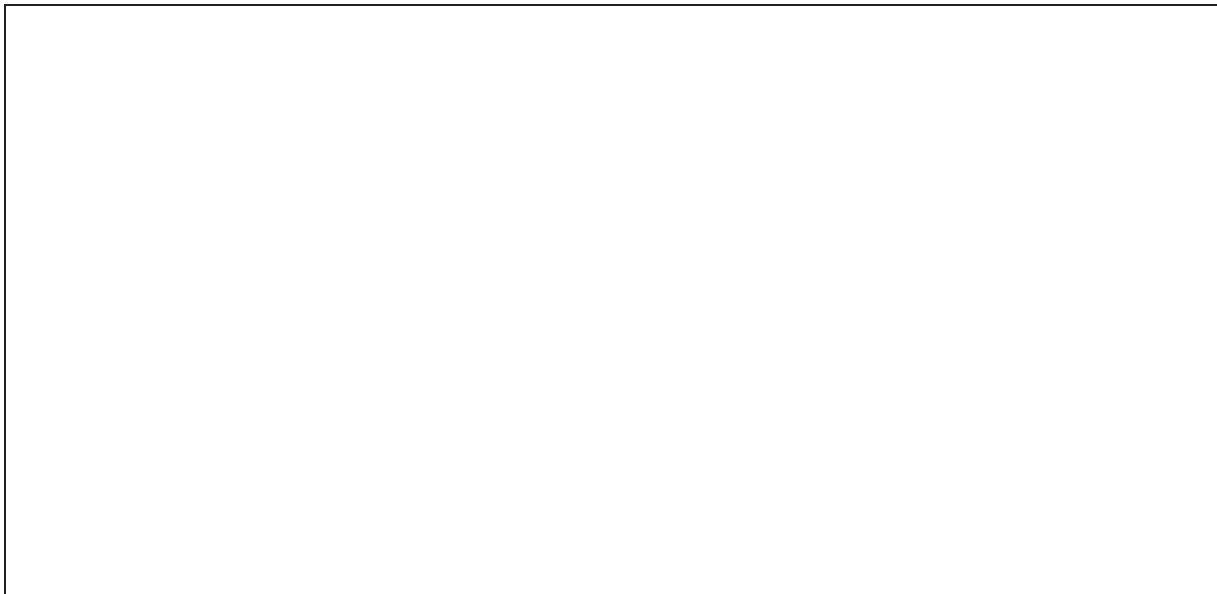
Continue to Short Answer questions on next page.

C. Short answer questions (2 x 3 → 6 %)

C1. Describe (qualitatively) the change in the phase (or, “delay”) ϕ and amplitude A of the annual temperature wave with increasing depth z into a soil.



C2. Consider an isothermal planet of radius R that has no atmosphere, whose solar constant is S_0 , and whose albedo is $a = 0.7$. Making reference to a statement of proportionality ($dT/dt \propto \dots$), explain the feedback that stabilizes the temperature T of this hypothetical planet.



Equations and Data.

- one full barb on the wind vector corresponds to 5 m s^{-1} , and a solid triangle means 25 m s^{-1} .
- $\frac{\Delta P}{\Delta z} = -\rho g$. ΔP [Pascals], the change in pressure as one ascends a distance Δz [m]; ρ [kg m^{-3}] the air density; $g = 9.81 \text{ [m s}^{-2}\text{]}$ acceleration due to gravity.
- $R = R^*/M$. Gives the specific gas constant R for a gas composed on molecules having molecular mass $M \text{ kg mol}^{-1}$, where $R^* = 8.3143 \text{ J mol}^{-1} \text{ K}^{-1}$ is the universal gas constant. To a first approximation the molecular mass of an *atom* is given by its atomic number, i.e. its place in the periodic table of the elements.
- $P = \rho R_d T_v$. The ideal gas law. P [Pascals], total pressure; ρ , [kg m^{-3}] the total density; T_v [Kelvin], the virtual temperature; and $R_d = 287 \text{ [J kg}^{-1} \text{ K}^{-1}\text{]}$, the specific gas constant for dry air.
- $\Delta q = c_v \Delta T + P \Delta \alpha$. The first law of thermodynamics, linking changes in the state of a sample of air: $\Delta q \text{ [J kg}^{-1}\text{]}$ is energy added to the system (zero for an adiabatic process, by definition), c_v is the specific heat capacity of the material at constant volume, P is the pressure and $\alpha \equiv 1/\rho$ is the specific volume (ρ being air density). The $c_v \Delta T$ term is the change in the *internal energy* of the system. For an ideal gas $c_v = (5/2)R$, so that for dry air $c_v \approx 720 \text{ J kg}^{-1} \text{ K}^{-1}$.

$$\frac{T}{T_1} = \left(\frac{P}{P_1} \right)^{R/c_p}$$

Poisson's law linking two states (P, T) and (P_1, T_1) of a sample of ideal gas, assuming the process connecting the two states is adiabatic ($R/c_p = 2/7 = 0.286$).

- $\frac{\Delta T}{\Delta z} = -\frac{g}{c_p}$. The dry adiabatic lapse rate, where g is the gravitational acceleration and $c_p \text{ [J kg}^{-1} \text{ K}^{-1}\text{]}$ is the specific heat at constant pressure. Our textbook defines the DALR as the *magnitude*, $\text{DALR} = |\Delta T/\Delta z|$, often rounded to $\text{DALR} = 1^\circ\text{C}/100 \text{ m}$.

$$Q = -k \frac{\Delta T}{\Delta x}$$

Fourier's law of conduction, giving the conductive heat flux density Q in direction x if there is a temperature gradient $\Delta T/\Delta x$ in a medium whose conductivity is $k \text{ W m}^{-1} \text{ K}^{-1}$. (The sign convention is that Q is positive for a flow of heat towards larger values of x .)

- $E = \epsilon \sigma T^4$

Stefan-Boltzmann law for a greybody. E [W m^{-2}], the emitted longwave energy flux density; ϵ , the emissivity of the surface (dimensionless); $\sigma = 5.67 \times 10^{-8}$ [$\text{W m}^{-2} \text{K}^{-4}$], the Stefan-Boltzmann constant; T [K], the surface temperature.

- $\lambda_{\max} = \frac{2897}{T}$

Wien's displacement law. λ_{\max} [μm], the wavelength at which the peak in the emission spectrum occurs; T [K], the temperature of the emitting surface.

- $Q^* = Q_H + Q_E + Q_G$. Surface energy balance on a reference plane at the base of the atmosphere. Fluxes normally in [W m^{-2}], but sometimes expressed as totals [J m^{-2}] over (e.g.) a day or a year. Q^* the net radiation, positive if directed towards the surface; Q_H, Q_E the sensible and latent heat fluxes, positive if directed from the surface towards the atmosphere; Q_G the 'soil' heat flux, positive if directed from the surface into ground/lake/ocean.

- $\alpha = 90^\circ - |\phi - \delta|$. Solar elevation α at local solar noon time, at a location having latitude ϕ (negative in the S. hemisphere), at the time of year when solar declination is δ (δ is negative during northern hemisphere winter, and its magnitude never exceeds 23.5°).

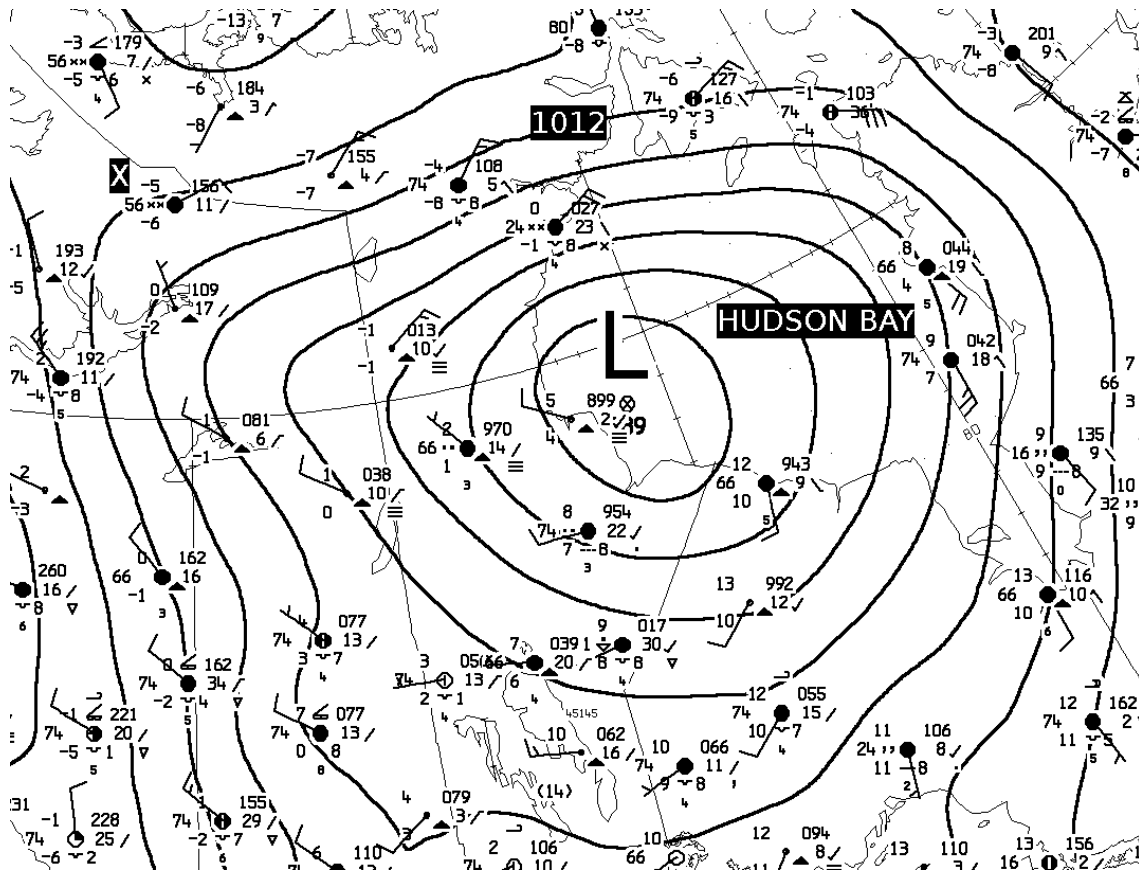


Figure 1: Environment Canada surface analysis (cropped) for 12 UTC Thurs. 2 October 2014.

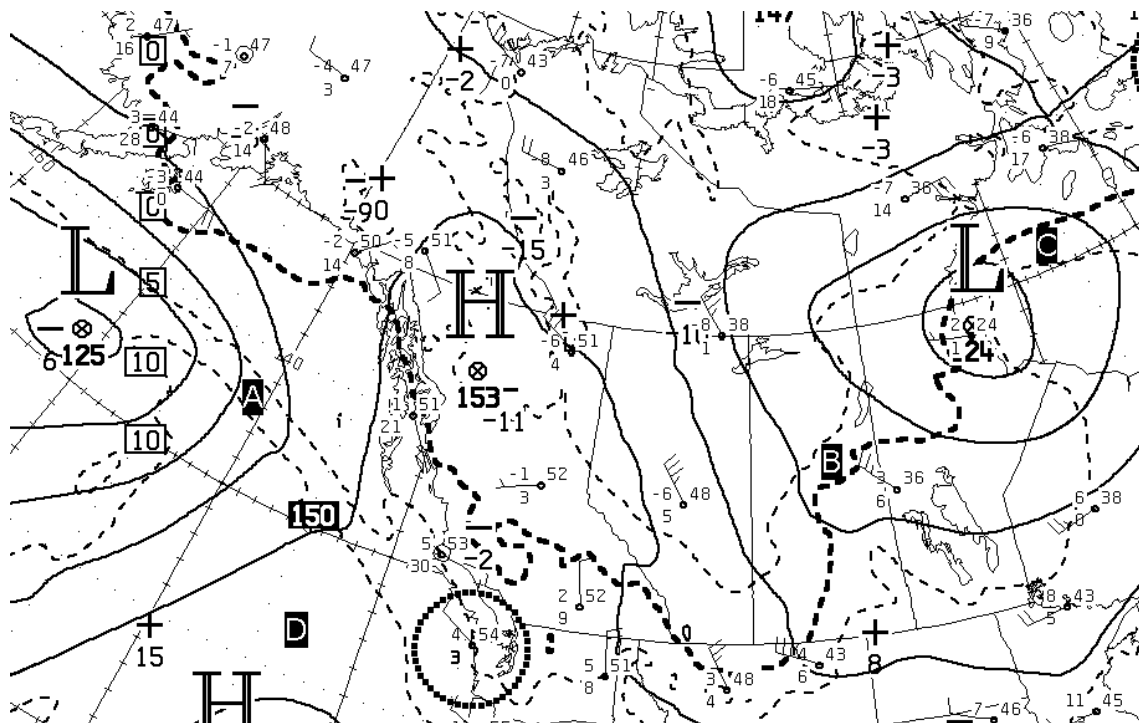


Figure 2: Environment Canada 850 hPa analysis (cropped) for 12 UTC Thurs. 2 October 2014.